

Chemical Crystallography and Structural Chemistry

VO 270063-1

Lecture N° 4 — 30^{th} March 2023

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Course Schedule

2 rd	March	Lecture Nº 1
16^{th}	March	Lecture Nº 3
30 st	March	Lecture Nº 4
27 th	April	Exercise Nº 2
$11^{\rm th}$	May	Exercise Nº 3
1^{st}	June	no lecture
22 nd	June	Exercise $N^{\underline{o}}$ 4

9^{th}	March	Lecture Nº 2
23 th	March	Exercise Nº 1
20 th	April	Lecture № 5
4^{th}	May	Lecture № 6
25 th	May	Lecture № 7
15^{th}	June	Lecture № 8
29 th	June	Lecture Nº 9



Previous Lecture

- Crystallographic symmetry oparations
- Crystallographic point groups and space groups
- Unit cell conventions
- Crystal systems and Bravais lattices

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1 Crystal Systems [1]



The 7 crystal systems [1]

Indepently from the content of the unit cell, the unit cell parameters $a,b,c,\alpha,\beta,\gamma$, can be classified according to regularity. These classifications result in the seven crystal systems

System		Conditions	Conditions
		lengths	angles
triclinic	а	none	none
monoclinic	m	none	$\alpha=\gamma=90^\circ$
trigonal	h	a = b = c	$\alpha=\beta=90^\circ$
			$\gamma = 120^\circ$
hexagonal	h	a = b	$\alpha=\beta=90^\circ$
			$\gamma = 120^\circ$
orthorhombic	0	none	$\alpha=\beta=\gamma=90^\circ$
tetragonal	t	a = b	$\alpha=\beta=\gamma=90^\circ$
cubic	с	a = b = c	$\alpha=\beta=\gamma=90^\circ$



The 7 crystal systems [1]

Indepently from the content of the unit cell, the unit cell parameters $a,b,c,\alpha,\beta,\gamma,$ can be classified according to regularity. These classifications result in the seven crystal systems





14 Bravais lattices [1]

7 crystal systems plus the requirement that the unit cell reflects the symmetry results in **14 Bravais lattices**:



Dots: Positions of symmetry elements



14 Bravais lattice — naming conventions

- lower case letters (a, m, h, o, t, c): crystal system (shape of cell).
- upper case letters (P, C, F, I, R): Symmetry. They present positions of symmetry elements within the unit cell in addition to the unit cell corners.
 - P primitiv, symmetry elements only at corners
 - **C** C-centred: Symmetry element at centre of C-faces (spanned by \vec{a} and \vec{b}); (correspondingly: A- and B)
 - F face centred (centred at each face)
 - I body centred: at the centre of the cell (German: innenzentriert)
 - ${\bf R}~{\bf r}$ hombohedral
- the (P, C, F, I, R) are the same as in the Hermann-Mauguin names for space groups (P2₁2₁2, C2/c, ...)



Crystal system \neq Space group

Do not draw conclusions on the space group from the shape of the unit cell: the space group **also** depends on the cell content.



orthorhombic box, non-symmetric space group



Positions and naming of unit cell axes: Conventions

- The origin (0,0,0) is placed at a position of high symmetry. If present, it is placed at a centre of inversion
- There are $6 \times 4 \times 2 = 48$ possibilities to name the axes as \vec{a} , \vec{b} , und \vec{c}
- Conventions:
 - 1. right-handed coordinate system
 - 2. highest possible symmetry
 - 3. small as possible volume of unit cell
 - $\textbf{4.} \ a \leq b \leq c$





- Despite these conventions, *some* space groups still have several possible axis settings
- For a single data set, the choice is arbitrary
- when two data sets or more are collected, care must be taken to index both consistently
- Modern programs automatically take care of consistent indexing.



Example: trigonal space group P3

!!! WARNING !!! SOLUTION MAY NOT BE UNIQUE.

UNIT_CELL_A-AXIS=	-140.027	-180.486	-28.163
UNIT_CELL_B-AXIS=	188.097	21.258	-130.933
UNIT_CELL_C-AXIS=	123.664	-120.609	158.073

UNIT_CELL_A-AXIS= 188.097 21.258 -130.933 UNIT_CELL_B-AXIS= -140.027 -180.486 -28.163 UNIT_CELL_C-AXIS= -123.664 120.609 -158.073

UNIT_CELL_A-AXIS= 48.070 -159.228 -159.096 UNIT_CELL_B-AXIS= -188.097 -21.258 130.933 UNIT_CELL_C-AXIS= -123.664 120.609 -158.073





- Single data set: choice of axis settings is irrelevant, each one is equally good
- Several data sets: with "wrong" settings, intensities do not match weil unterschiedlich indiziert
- either from the same crystal or different crystal
- re-indexing of second data set required in order to match intensities of the first one \rightarrow Reindexing of the coordinate system required
- Worst case: testing of all possibilities
- automated in modern software



3 Symmetry and X-ray data



Symmetry and X-ray data

Overview

- Effect of symmetry on X-ray data
- Friedel's law
- Laue groups: point groups in reciprocal space



Note: Calculation of reflex intensities

Once the chemical composition *i.e. atom types and their positions*, of the asymmetric unit is know, one can calculate the electron density $\rho(x,y,z)$. The intensity of every reflection can be calculated from the electron density $\rho(x,y,z)^1$

$$I(hkl) = \left| \int_{V(\mathbf{u.c.})} \rho(x,y,z) e^{2\pi i (hx+ky+lz)} d^3x \right|^2$$

¹International Tables Volume B, Ch. 1.2



Symmetry in reciprocal space

Reminder: A symmetry operation (R, \vec{t}) leaves the entire crystal, and in particular the unit cell, unchanged, *i.e.* for every point in the unit cell:

$$\rho(R\vec{x} + \vec{t}) = \rho(\vec{x})``u.c." = R \cdot ``u.c." + \vec{t}$$

This translates towards the intensities:

$$\Rightarrow I(hkl) = \left| \int_{V(\mathbf{u.c.})} \rho(x, y, z) e^{2\pi i (hx + ky + lz)} d^3x \right|^2$$
$$= \left| \int_{V(\mathbf{u.c.})} \rho(R\vec{x} + \vec{t}) e^{2\pi i (\vec{h}^T (R\vec{x} + \vec{t}))} d^3 (R\vec{x} + \vec{t}) \right|^2$$
$$= I(R^T \vec{h})$$

(The last step include quite some calculations)

This means: The intensities of the two reflections $(\boldsymbol{h},\boldsymbol{k},\boldsymbol{l})$ and $R(\boldsymbol{h},\boldsymbol{k},\boldsymbol{l})$ are identical.



Symmetry in reciprocal space

For every symmetry operation (R, \vec{t}) of the space group of a crystal:

$$I(\vec{h}) = I(R^T \vec{h}) = I(R^T R^T \vec{h}) = \dots$$

- Crystal symmetry also means symmetry for the reflections
- Translational symmetry is "dropped" in reciprocal space
- Translational symmetry results in a phase shift, that cannot be measured.



Friedel's Law

Even without any symmetry, *i.e.* in all space groups including *P*1:

 $I(h,k,l)=I(\bar{h},\bar{k},\bar{l})$

Reciprocal space always has a centre of inversion. This is called Friedel's law²

²Georges Friedel, 1865–1933



Excursus: Mathematics for Friedel's Law

$$\begin{split} I(\bar{h}\bar{k}\bar{l}) &= \left| \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{2\pi i (\bar{h}x + \bar{k}y + \bar{l}z)} d^3x \right|^2 \\ &= \left| \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{-2\pi i (hx + ky + lz)} d^3x \right|^2 \\ &= \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{-2\pi i (hx + ky + lz)} d^3x * \left(\int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{-2\pi i (hx + ky + lz)} d^3x \right)^* \\ &= \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{-2\pi i (hx + ky + lz)} d^3x * \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{2\pi i (hx + ky + lz)} d^3x \\ &= \left| \int_{V(\mathsf{E},\mathsf{Z},)} \rho(x,y,z) e^{2\pi i (hx + ky + lz)} d^3x \right|^2 \\ &= I(hkl) \end{split}$$



Measurement of Friedel pairs

Two reflections I(hkl) and $I(\bar{h}\bar{k}\bar{l})$ form a **Friedel pair**. They can be measured *e.g.* by rotation the crystal by 180°





Phase difference of Friedel pairs



$$\Delta_{\text{left}} = -\Delta_{\text{right}}$$
$$\Rightarrow \delta_{\text{left}} = -\delta_{\text{right}} \qquad (\delta = 2\pi\Delta/\lambda)$$

negative phase, same path difference: $I(hkl) = I(\bar{h}\bar{k}\bar{l})$ and $\Phi(hkl) = -\Phi(\bar{h}\bar{k}\bar{l})$

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Laue groups

- Crystals belong to one of 230 space groups
- The diffraction pattern contains the symmetry of the space group, but:
 - 1. no translational component
 - 2. always centre of inversion (Friedel's law)
- This results in a subset of groups: the 11 Laue groups.



11 Laue groups

Crystal system	Laue group	point group
Triclinic	Ī	1, Ī
Monoclinic	2/m	2, m, 2/m
Orthorhombic	mmm	222, mm2, mmm
Tetragonal	4/m	4, $\overline{4}$, 4/m
	4/mmm	422, 4mm, $ar{4}2m$, 4/mmm
Trigonal	$\bar{3}$	3, <u>3</u>
	$\bar{3}m$	32, 3m, $ar{3}m$
Hexagonal	6/m	6, ō̄, 6/m
	6/mmm	622, 6mm, $ar{6}m2$, 6/mmm
Cubic	$m\bar{3}$	23, $mar{3}$
	$m\bar{3}m$	432, $\bar{4}3m$, $m\bar{3}m$

http://de.wikipedia.org/wiki/Lauegruppe



Extinctions

- Extinctions are reflections with systematically zero intensity.
- Extinctions occur in non-primitive Bravais lattices or in the presence of symmetry elements with translational part, *i.e.* glide planes and screw axes.
- Extinctions are important for space group determination, *e.g.* to differentiate between P4 and $P4_3$.
- The "Int. Tabl. Vol. A" list the "general reflection conditions", *i.e.* those reflections with non-zero intensity:

 $P3_1$ (No. #144) 00l: l = 3n

- only if $(hkl)=\ldots(00-6),\;(00-3),\;(003),\;\ldots$ Intensities $\neq 0;$
- reflections (001), (002), (004), (005), ..., should have zero intensity



Example: Extinctions in $P4_32_12$

- Reflection condition: h00: h = 2n and 00l: l = 4n
- Only reflections with k=0, l=0 and h even should have non-zero intensity
- All reflections with k = 0, l = 0 and h odd should have zero intensity
- Example: Lysozyme data set





Example: Extinctions in $P4_32_12$

Output from the program xprep for Lysozyme data:

System	atic a	bsence	excep	tions:					
41/43	42	n	-b-	-c-	-n-	-21-	c		
N	5	3	3490	2365	2359	2358	89	1549	
N I>3s	0	0	2188	1256	1361	1275	0	931	
<i></i>	0.9	0.9	97.0	64.6	80.7	65.4	0.8	85.0	
<i s=""></i>	0.2	0.2	8.2	6.7	7.3	6.6	0.4	7.5	
Option	Space	e Grou	p No.	Туре	Axes	CSD	R(sym) N(eq)	Syst.
Abs.	CFOM								
[A] P4	(1)2(1)2	# 92	chiral	. 1	245	0.067	140065	0.4 /
6.6	1.99								
[B] P4	(3)2(1)2	# 96	chiral	. 1	245	0.067	140065	0.4 /
6.6	1.99								

Based on the data alone, we cannot distinguish between $P4_12_12$ and $P4_32_12$ (due to Friedel's law). One has to solve the structure and see which one makes sense. For protein structures: the right space group will consist of L-amino acids, the wrong one of D-amino acids.



Summary: Space group determination

- Crystal symmetry is present in reflection data intensities
- No translational symmetry in reciprocal space
- Only one of the 11 Laue groups can be determined from symmetry in intensities
- Systematic absences are important to distinguish between space groups within a Laue group



4 Experimental procedure: from data collection to structure

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From Data Collection to Structure



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Chemical Crystallography



Data Collection

			-
			-

several GB



0 0 1 4.0 1.0 0 0 -2 1'257.0 35.5 0 0 -2 1'600.0 42.7

several files, 100's $\ensuremath{\mathsf{MB}}$



Data Scaling

0	0	-1	2.8	0.55
0	0	1	3.8	0.63
0	0	-2	1'432.0	95.7
0	0	-2	1'282.0	85.9

1 "hkl"-file, 50MB





How to collect good data

Data resolution

Data completeness

Data multiplicity

Data quality



Data resolution

- highest possible resolution: $d_{\min} > \lambda/2$
- good crystals diffract better than this limit (instrumentation limited resolution)
- bad crystals may not reach this resolution (sample limited resolution)
- NB: low resolution are also important





Data completeness

- Only reflections with resolution sphere can be collected (radius in reciprocal space: $r=2./\lambda$)
- Laue equations permit to compute their location, and how to rotate the crystal into an appropriate position.
- Single rotation axis: torus shape of reflections.



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Data multiplicity

- Intensity $\left(I(hkl) \text{ of the identical reflection } (hkl) \text{ can be measured multiple times} \right.$
- Intensities I(hkl) and $I^\prime(R(hkl))$ symmetry equivalent reflections increase multiplicity
- Measuring on different positions on detector: reduction of systematic errors
- increasing multiplicity reduces uncertaintig $\sigma(I)$



Data quality

- "Diffraction data acquisition is the final experimental stage of the crystal structure analysis." (Z. Dauter, [2])
- Ensure proper data collection strategy (desired resolution, multiplicity; adjust detector distance to unit cell parameters, etc.)
- Crystal quality can be the limiting factor



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Steps of data integration: initial diffraction images



- typically four short runs:
- $2 \times 10^{\circ} 15^{\circ}$: $0^{\circ} 10^{\circ}$ and $90^{\circ} 100^{\circ}$
- two detector settings each (low to mid resolution and mid to high resolution)



Steps of data integration: peak finding



- simple (non-crystallographic) pattern recognition algorithm: peak above background
- main parameter: desired peak strength (I/sigma)



Steps of data integration: finding unit cell



- known instrument geometry: project peaks into reciprocal space
- find a lattice basis



Steps of data integration: finding unit cell





Data Integration





Data Integration

Data integration comprises

- 1. Indexing: Determination of unit cell dimensions, orientation of the crystal, point group
- 2. Extraction of spot intensities from detector images.
- 3. Optimisation of experimental parameters



Programs for data integration (incomplete)

- Saint Licensed by Bruker AXS. Specific to Bruker programs. Good for data from twinned crystals. Derived from XDS.
- **X-Area** Software suite distributed with Stoe X-ray diffractometers. Free to use. Very good interactive indexing routine, good for twinned data.
- XDS Free for non-commercial users (http://xds.mpimf-heidelberg.mpg. de). Supports nearly all detector formats, very well documented. Very fast.
- **DIALS** Free for non-commercial users. Very active development. (https: //dials.diamond.ac.uk/)
- **EVAL Suite** Free for non-commercial users (http://www.crystal.chem. uu.nl/distr/eval). Can integrate *e.g.* incommensurate crystals

Crysalis Pro Licensed by Rigaku (https://www.rigaku.com/products/ smc/crysalis)



iMosflm Free for non-commercial users, distributed with CCP4 (http://
www.ccp4.ac.uk)

HKL3000 Very good visualisation GUI for fine-tuning of parameters. very popular in the US (https://hkl-xray.com/)



Indexing



"Ewald sphere backwards:"

- find 200–1000 strong spots
- backtransform into reciprocal space (Laue equations)
- find a lattice and a suitable basis
- basis for reciprocal lattice corresponds to reduced unit cell constants



Indexing



Possible reasons for indexing problems:

- Incorrect parameters: detector distance, direction of rotation, wavelength, especially at synchrotrons
- Too few reflections
- Distorted spots (lattice defects, unfocused beam)
- Alien spots (ice, metal, contaminant)
- multiple lattices twins



References

- [1] W. Massa. *Crystal Structure Determination*. Springer Verlag Berlin, Heidelberg, 2002.
- [2] Zbigniew Dauter. 'Collection of X-Ray Diffraction Data from Macromolecular Crystals'. In: Protein Crystallography: Methods and Protocols. Ed. by Alexander Wlodawer, Zbigniew Dauter and Mariusz Jaskolski. New York, NY: Springer New York, 2017, pp. 165–184. DOI: 10.1007/978-1-4939-7000-1_7.