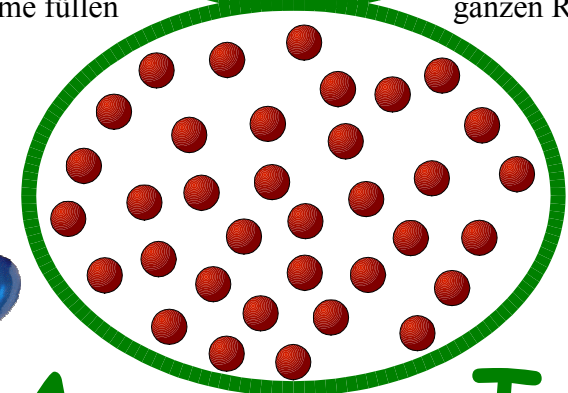
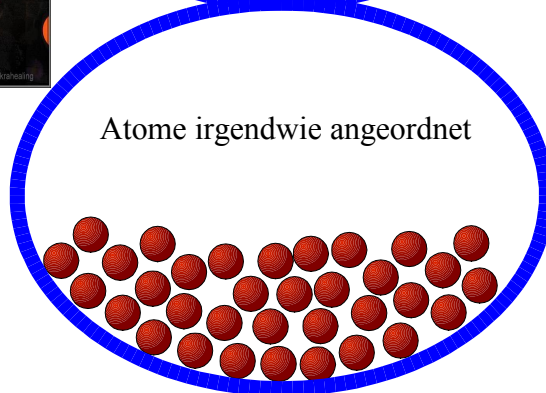
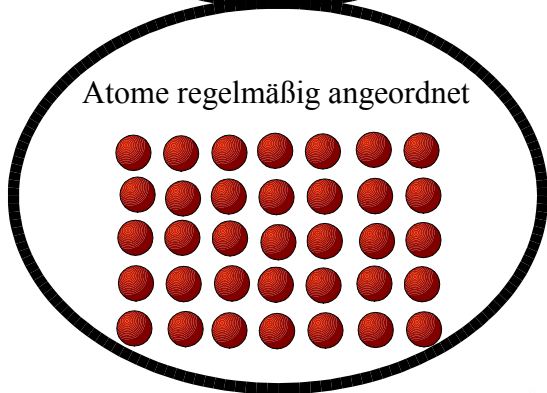
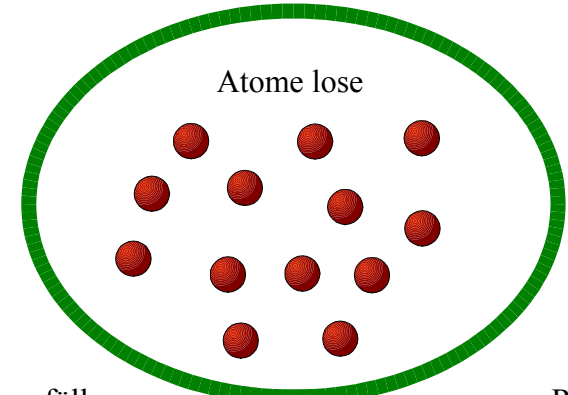
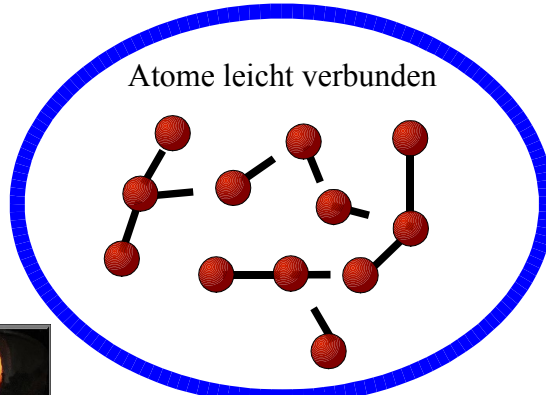
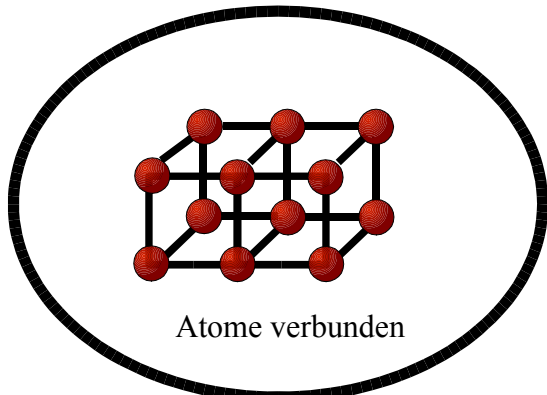


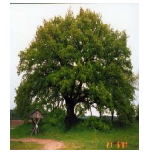
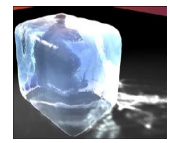
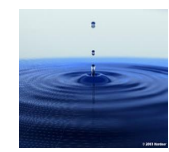
Alle Materie besteht aus Atomen !



FEST

FLÜSSIG

GASFÖRMIG



Martoks Fotogalerie

Atome

112 Sorten (Elemente)

Unterschiede: **Gewicht (Größe)**
Eigenschaften

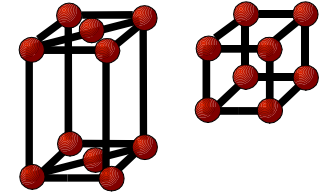


Beispiele: Kohlenstoff = C
Sauerstoff = O
Wasserstoff = H
Eisen = Fe
Gold = Au

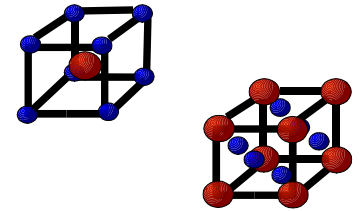
Die Chemie beschäftigt sich mit der Bindung
und den Eigenschaften der Atome

Alle Materie besteht aus Atomen !

- Verschiedener **Aufbau**



- **Mischung** der Sorten



- **Temperatur**

